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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/810,081	03/25/2004	Robert C. West	Q199-US1	8213
Quallion LLC 7590 12/23/2008				
P.O. Box 923127				
Sylmar, CA 91392-3127				
EXAMINER				
BEST, ZACHARY P				
ART UNIT		PAPER NUMBER		
1795				
MAIL DATE		DELIVERY MODE		
12/23/2008		PAPER		

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Office Action Summary

Application No.

10/810,081

Applicant(s)

WEST ET AL.

Examiner

Zachary Best

Art Unit

1795

Period for Reply -- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 08 October 2008.
- 2a) ☐ This action is **FINAL**. 2b) ☒ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 1-7, 38-49, 51-90 is/are pending in the application.
- 4a) Of the above claim(s) 38-48 and 55-65 is/are withdrawn from consideration.
- 5) ☐ Claim(s) _____ is/are allowed.
- 6) ☒ Claim(s) 1-7, 49, 51-54, 66-90 is/are rejected.
- 7) ☐ Claim(s) _____ is/are objected to.
- 8) ☐ Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on _____ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

- 12) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All b) ☐ Some * c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
 2. ☐ Certified copies of the priority documents have been received in Application No. _____.
 3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

- 1) ☒ Notice of References Cited (PTO-892)
- 2) ☐ Notice of Draftsperson's Patent Drawing Review (PTO-948)
- 3) ☐ Information Disclosure Statement(s) (PTO/SB/888)
Paper No(s)/Mail Date _____
- 4) ☐ Interview Summary (PTO-413)
Paper No(s)/Mail Date _____
- 5) ☐ Notice of Informal Patent Application
- 6) ☐ Other: _____

**ELECTROLYTE INCLUDING POLYSILOXANE
WITH CYCLIC CARBONATE GROUPS**

Examiner: Z. Best S.N. 10/810,081 Art Unit: 1795 December 17, 2008

DETAILED ACTION

1. Applicant's amendment and request for reconsideration filed on October 8, 2008 was received. Claim 2 was amended. Claims 8-37 and 50 were cancelled. Claims 38-48 and 55-65 were withdrawn. Claims 1-7, 49, 51-54, and 66-90 are currently pending examination.
2. The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.

Specification

3. The objection to the specification is withdrawn because the specification was amended.

Claim Objections

4. The objection to Claims 50 and 82 are withdrawn because Claim 50 was cancelled.

Claim Rejections - 35 USC § 103

5. The rejections under 35 U.S.C. 103(a) of Claims 1-6, 49-52, 66-69, 71-74, 76, 78, 81-84, 86, and 88 as being unpatentable over Lersch et al. (US 5,606,077 A) in view of Kang et

al. (US 6,783,897 B2), Claims 7, 53, 70, 75, and 85 as being unpatentable over Lersch et al. in view of Kang et al., and further in view of Hosoya (US 2001/0036579 A1), Claims 54, 77, and 87 as being unpatentable over Lersch et al. in view of Kang et al., and further in view of Carlson et al. (US 2002/0092155 A1), and Claims 79-80 and 89-90 as being unpatentable over Lersch et al. in view of Kang et al., and further in view of Lee (US 6,680,147 B2) are withdrawn because Applicant's arguments are persuasive.

6. Claims 1-6, 49, 51-52, 66-69, 71-74, 76, 78, 81-84, 86, and 88 are rejected under 35 U.S.C. 103(a) as being unpatentable over Spiegel et al. (US 6,447,952 B1) in view of Kang et al. (6,783,897 B2).

Regarding Claims 1, 49, 71, 76, 81, and 86, Spiegel et al. teach a battery (electrochemical device, abstract) comprising a polymer electrolyte that is elastomeric (solid) including a polysiloxane having one or more backbone silicons linked to a first side chain and one or more backbone silicons linked to a second side chain (formulas I and II, col. 3, lines 32-44), wherein one of aforesaid chains (second side chain) comprises a cyclic carbonate group (col. 4, lines 12-18). However, Spiegel et al. fail to teach the other aforesaid chain (first side chain) comprises a polyalkylene oxide group.

Kang et al. teach an electrolyte having a polysiloxane polymer as the backbone (abstract), wherein the backbone is crosslinked with polyalkylene oxide groups (col. 3, lines 3-22) because the polyalkylene oxide group provides an further increase the compatibility with a plasticizer and improve ionic conductivity (col. 4, lines 31-46). Therefore, it would

have been obvious to one having ordinary skill in the art to crosslink the electrolyte comprising polysiloxane of Spiegel et al. with polyalkylene oxide groups because Kang et al. teach a polysiloxane crosslinked with polyalkylene oxide groups provides a further increase the compatibility with a plasticizer and improve ionic conductivity.

Regarding Claim 2, Spiegel et al. teach each of the non-terminal silicons in the backbone of the polysiloxane are crosslinked (formulas I and II, col. 3, lines 32-44).

Regarding Claims 3 and 66, Spiegel et al. teach the polysiloxane excludes Si-H groups (col. 4, lines 4-9).

Regarding Claims 4, 67, 72, and 82, Spiegel et al. teach a spacer between the cyclic carbonate group and the backbone comprising CH₂ groups (col. 3, lines 46-59), and Kang et al. teach a spacer between the polyalkylene oxide groups comprising CH₂ groups (cols. 7-8, lines 22-62).

Regarding Claims 5, 51, 68, 73, and 83, Spiegel et al. teach the polysiloxane having a structure according to the general formulas (claims 3, 6, 23-24, and 28).

Regarding Claims 6, 52, 69, 74, and 84, Spiegel et al. teach the average molecular weight for the polysiloxane is 400-5000 (col. 5, lines 6-11).

Regarding Claims 78 and 88, Spiegel et al. teaches that the polymer electrolyte contains modification agents (col. 2, line 62 - col. 3, line 13), and Kang et al. teach a network polymer (plasticizer) interacts with the polysiloxane so as to form an interpenetrating network (col. 8, lines 4-64).

7. Claims 7, 53, 70, 75 and 85 rejected under 35 U.S.C. 103(a) as being unpatentable over Spiegel et al. in view of Kang et al. as applied to Claims 1-6, 49, 51-52, 66-69, 71-74, 76, 78, 81-84, 86, and 88 above, and further in view of Hosoya (US 2001/0036579 A1).

Spiegel et al. in view of Kang et al. teach the electrolyte as recited in paragraph 6. However, Spiegel et al. in view of Kang et al. fail to teach the ratio between the molar concentration of the active oxygens in the electrolyte and the molar concentration of lithium ions in the electrolyte is 5-50.

Hosoya teach a lithium-ion battery comprising lithium ions in an electrolyte (par. 45), wherein the active oxygen should remain as small in concentration as possible because otherwise the active oxygen will begin to decompose the electrolyte (pars. 9-11). Therefore, it would have been obvious to one having ordinary skill in the art at the time the invention was made to create the electrolyte of Spiegel et al. in view of Kang et al. wherein the active oxygen concentration was kept as low as possible because Hosoya teach resultant electrolyte decomposition in a lithium ion battery when the active oxygen concentration increases. Discovery of an optimum value of a result effective variable involves only routine skill in the art. In re Boesch, 617 F.2d 272 (CCPA 1980).

8. Claims 54, 77, and 87 are rejected under 35 U.S.C. 103(a) as being unpatentable over Spiegel et al. in view of Kang et al. as applied to Claims 1-6, 49, 51-52, 66-69, 71-74, 76, 78, 81-84, 86, and 88 above, and further in view of Carlson et al. (US 2002/0092155 A1).

Spiegel et al. in view of Kang et al. teach the electrolyte as recited in paragraph 6. However, Spiegel et al. in view of Kang et al. fail to teach the solid polymer includes one or more components selected from the group consisting of polyacrylonitrile, polymethyl methacrylate, polyvinylidene fluoride, polyvinylidene fluoride-co-hexafluoropropylene, polystyrene, polyvinyl chloride, polyalkyl methacrylate, polyalkyl acrylate, styrene butadiene rubber, polyvinyl acetate, and polyethylene oxide.

Regarding Claims 54, 77, and 87, Carlson et al. teach an lithium electrochemical cell (par. 76) comprising an electrolyte, which may be a gel or solid, comprising polysiloxane (pars. 72-73), wherein said electrolyte may be a blend of polysiloxane and polyacrylonitrile (pars. 72-73). Therefore, it would have been obvious to one having ordinary skill in the art at the time the invention was made to create the electrolyte of Spiegel et al. in view of Kang et al. wherein the electrolyte is a gel or a solid polymer blend of polysiloxane and polyacrylonitrile because Carlson et al. teach functional equivalency of a gel or solid polymer electrolyte comprising polysiloxane and functional equivalency of a polysiloxane polymer electrolyte and a polysiloxane-polyacrylonitrile polymer electrolyte blend.

9. Claims 79, 80, 89, and 90 are rejected under 35 U.S.C. 103(a) as being unpatentable over Lersch et al. in view of Kang et al. as applied to Claims 1-6, 49, 51-52, 66-69, 71-74, 76, 78, 81-84, 86, and 88 above, and further in view of Lee (US 6,680,147 B2).

Spiegel et al. in view of Kang et al. teach the electrolyte as recited in paragraph 6. However, Spiegel et al. in view of Kang et al. fail to teach said network polymer includes a polymethacrylate or polyacrylate.

Lee teaches a lithium battery comprising a polysiloxane electrolyte (abstract), wherein the polysiloxane is combined with a dimethacrylate polymer (network polymer) (col. 8, lines 63-67). It would be advantageous to use the dimethacrylate polymer as a network polymer in the electrolyte because the polymeric electrolyte composition can effectively suppress swelling of the battery due to electrolytic solution (col. 2, lines 30-33). Therefore, it would have been obvious to one having ordinary skill in the art at the time the invention was made to create the electrolyte of Spiegel et al. in view of Kang et al., wherein the network polymer is a dimethacrylate polymer because Lee teaches the electrolyte can effectively suppress swelling of the battery due to electrolytic solution.

Response to Arguments

10. Applicant's arguments have been fully considered but are moot in view of the new ground(s) of rejection.

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Zachary Best whose telephone number is (571) 270-3963. The examiner can normally be reached on Monday to Thursday, 7:30 - 5:00 (EST).

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Dah-Wei Yuan can be reached on (571) 272-1295. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

zpb

/Dah-Wei D. Yuan/
Supervisory Patent Examiner, Art Unit 1795